



# Development and Application of CMAQ-MADRID-Mercury

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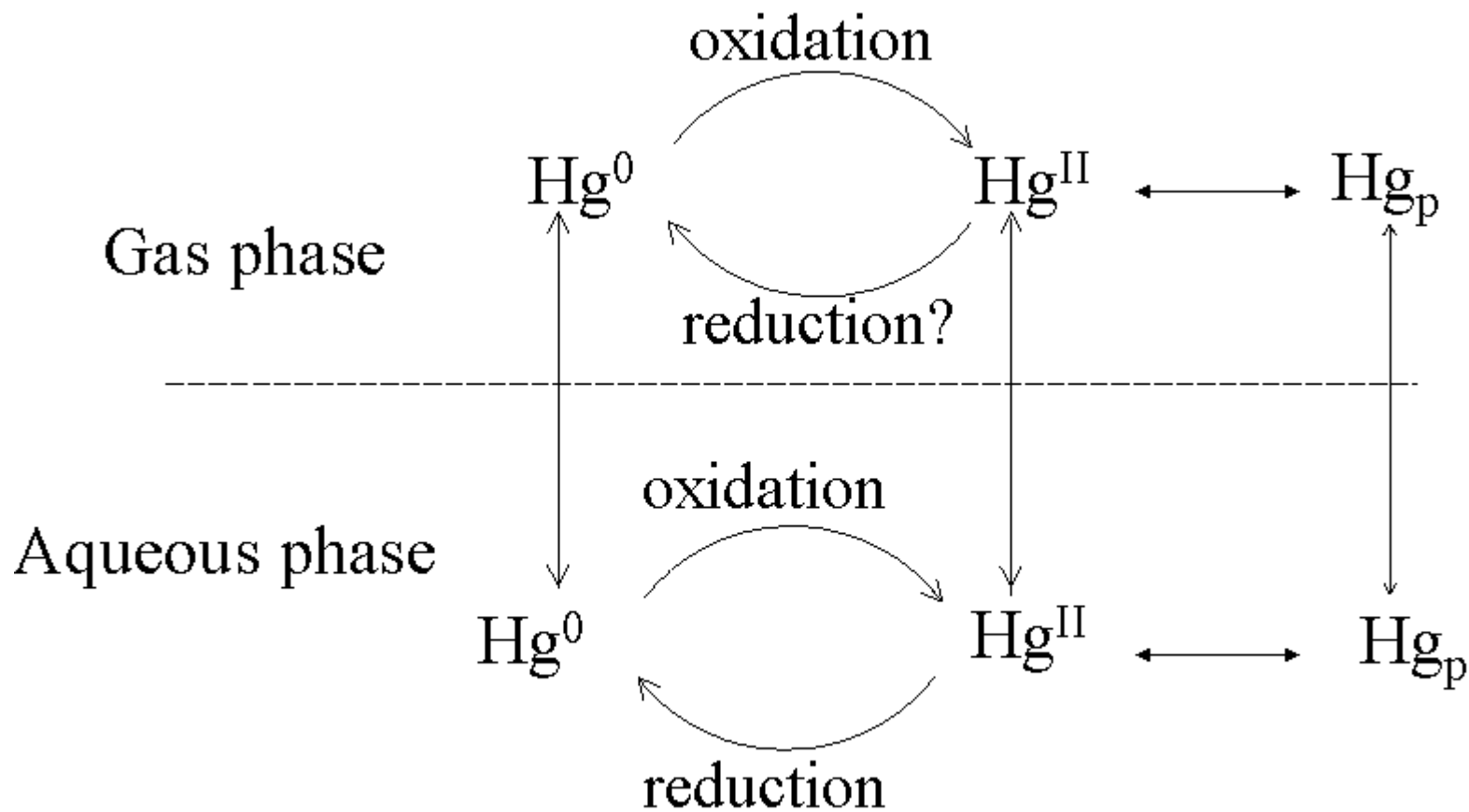
AER  
San Ramon, CA

CMAS Workshop 2005  
Chapel Hill, NC

# Atmospheric Mercury

- Mercury is present mostly as three “species” in the atmosphere
  - Elemental mercury
    - $\text{Hg}^0$
  - Divalent gaseous mercury
    - $\text{HgCl}_2$ ,  $\text{Hg}(\text{OH})_2$ ,  $\text{HgO}$ , etc.
    - referred to collectively as  $\text{Hg}^{\text{II}}$  or reactive gaseous mercury (RGM)
  - Particulate-bound mercury:
    - $\text{Hg}^{\text{II}}$  or  $\text{Hg}^0$  adsorbed on PM
    - mostly divalent
    - referred to collectively as  $\text{Hg}_p$

# Atmospheric Chemistry of Mercury



# Atmospheric Deposition of Mercury

- $\text{Hg}^0$  is not very soluble and has a low dry deposition velocity ( $<0.1$  cm/s)
- $\text{Hg}^{\text{II}}$  is very soluble and adsorbs readily on surfaces: it is rapidly removed by wet and dry deposition
- $\text{Hg}_p$  is mostly in the fine particle range and will remain in the atmosphere for several days in the absence of precipitation

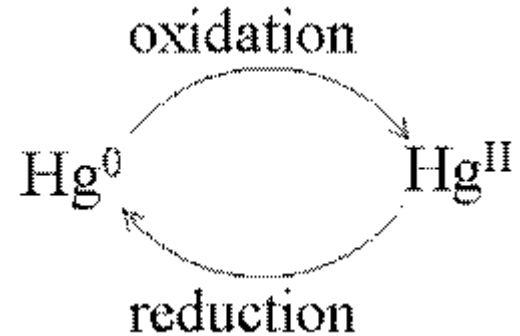
# Atmospheric Half-lives of Hg Species

- Hg(0)
  - Chemical oxidation: ~ 2-3 months on average\*
  - Dry deposition: ~ 3 months in the boundary layer
- Hg(II) or RGM
  - Chemical reduction: fast in presence of clouds
  - Dry deposition: ~ few hours-2 days in the boundary layer
  - Wet deposition: fast
- Hg(p)
  - Dry deposition: ~ 1 week in the boundary layer
  - Wet deposition: relatively fast

\* excluding Arctic and marine boundary layer chemistry

# Atmospheric Half-lives of Hg Species

Hg(0) and Hg(II) may undergo several red-ox cycles before Hg(II) is removed via dry or wet deposition



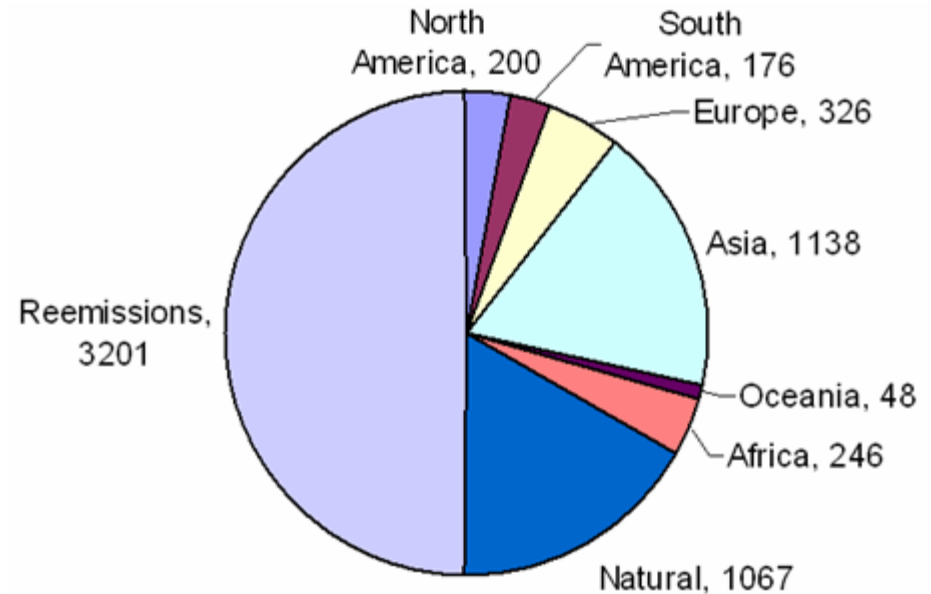
Atmospheric Hg has (currently) a half-life estimated to be ~ 10 months (lifetime of ~ 1.2 years)



**=> Hg is a global pollutant**

# Global Mercury Emissions

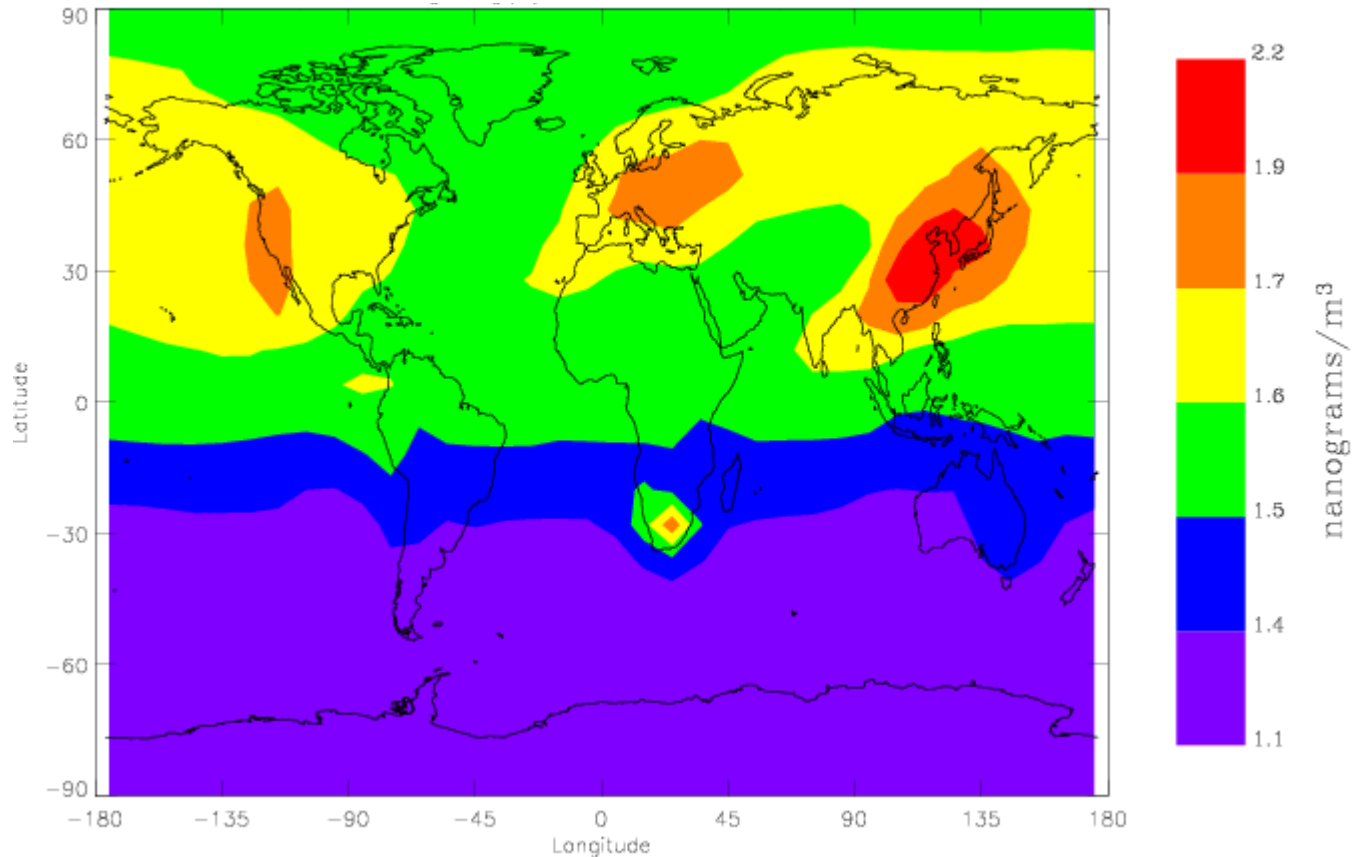
- Global emissions total about 6000 Mg/y with an uncertainty of a factor of 2
- Current emissions are about 3 times pre-industrial emissions
- Re-emissions of previously deposited mercury are a significant fraction of total emissions (30 to 50%)



**Annual Hg emissions  
(1999 estimate)**

# Global Simulation

## Annual surface concentrations of Hg(0)



- Good agreement with data (Seigneur et al., *ES&T*, 38, 555, 2004)
- Provides boundary conditions of mercury for the continental/regional model



# “One-atmosphere” Model of Atmospheric Mercury for Continental/Regional Scales

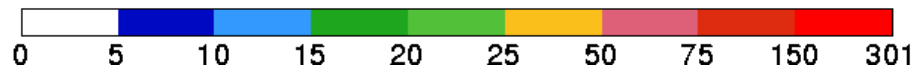
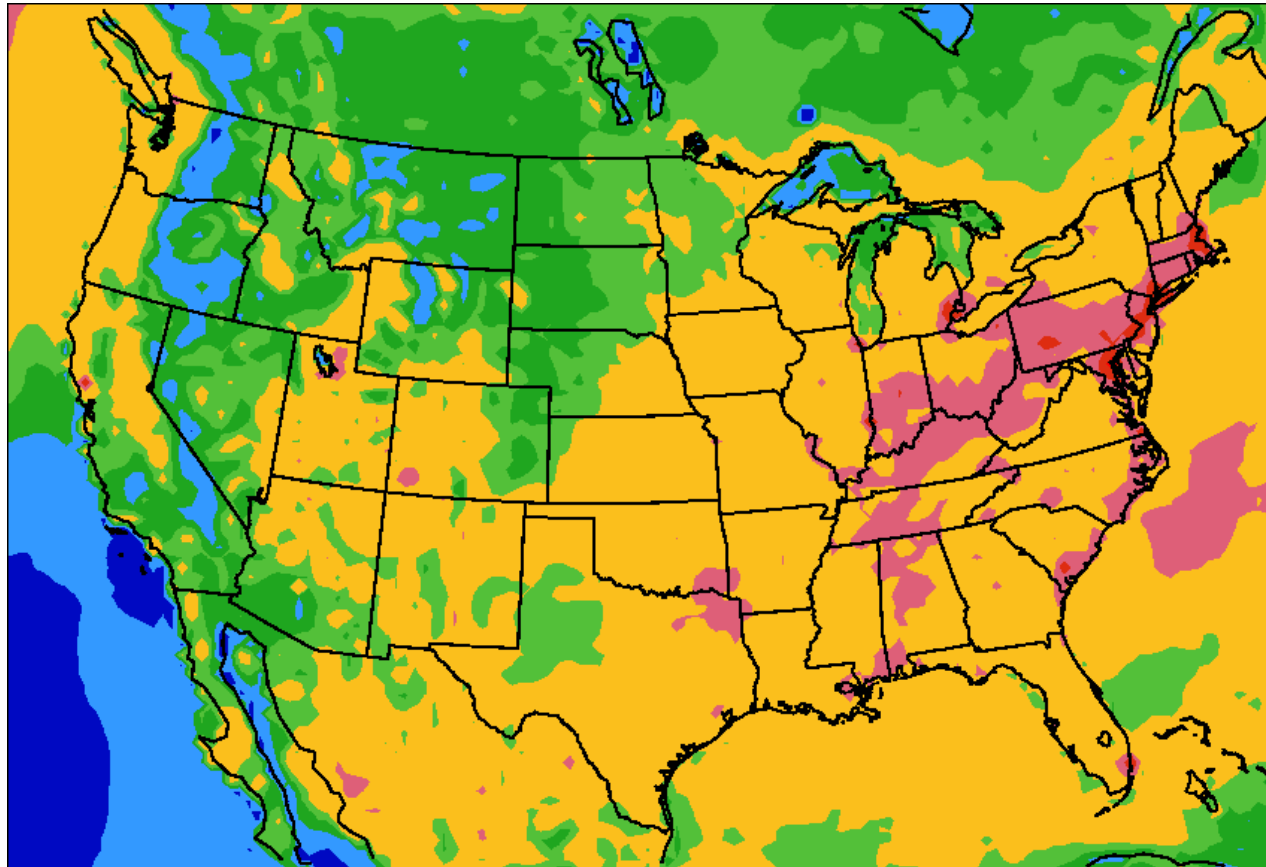
- Single state-of-the-science model to simulate ozone, PM and mercury
- EPA CMAQ as host model (version 4.4)
- State-of-the-science treatment of PM (MADRID)
- Advanced treatment of plumes for ozone and PM (APT)
- Incorporation of mercury processes (AER chemistry and removal processes)

# “One-atmosphere” Mercury Model

- CMAQ-MADRID-Hg is currently operational
  - Initial application to the continental U.S. for 1996
  - Comparison with MDN data completed
  - Available in the Model Download section at <http://www.cmascenter.org> (CMAQ-MADRID 2004)



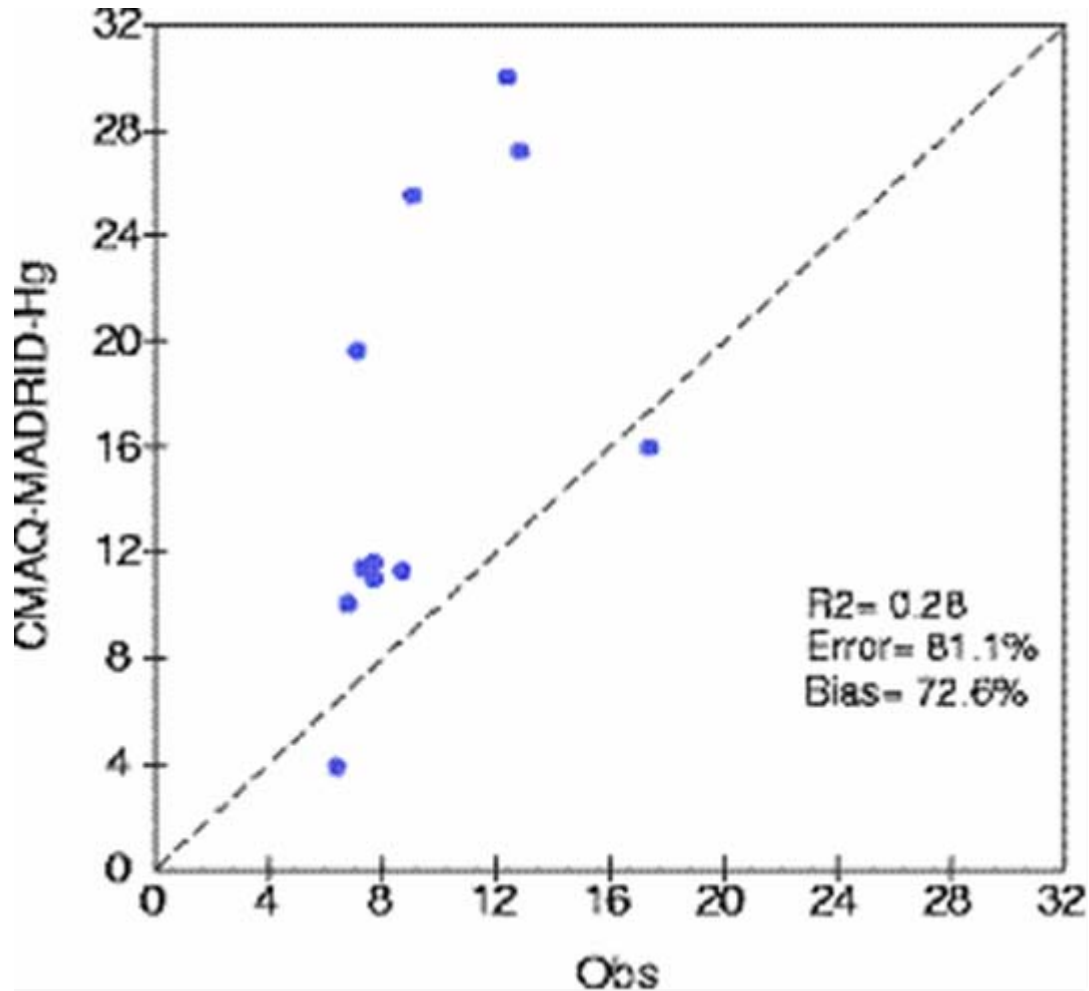
# Total Hg Deposition Simulated with CMAQ-MADRID-Hg for 1996 ( $\mu\text{g}/\text{m}^2\text{-yr}$ )





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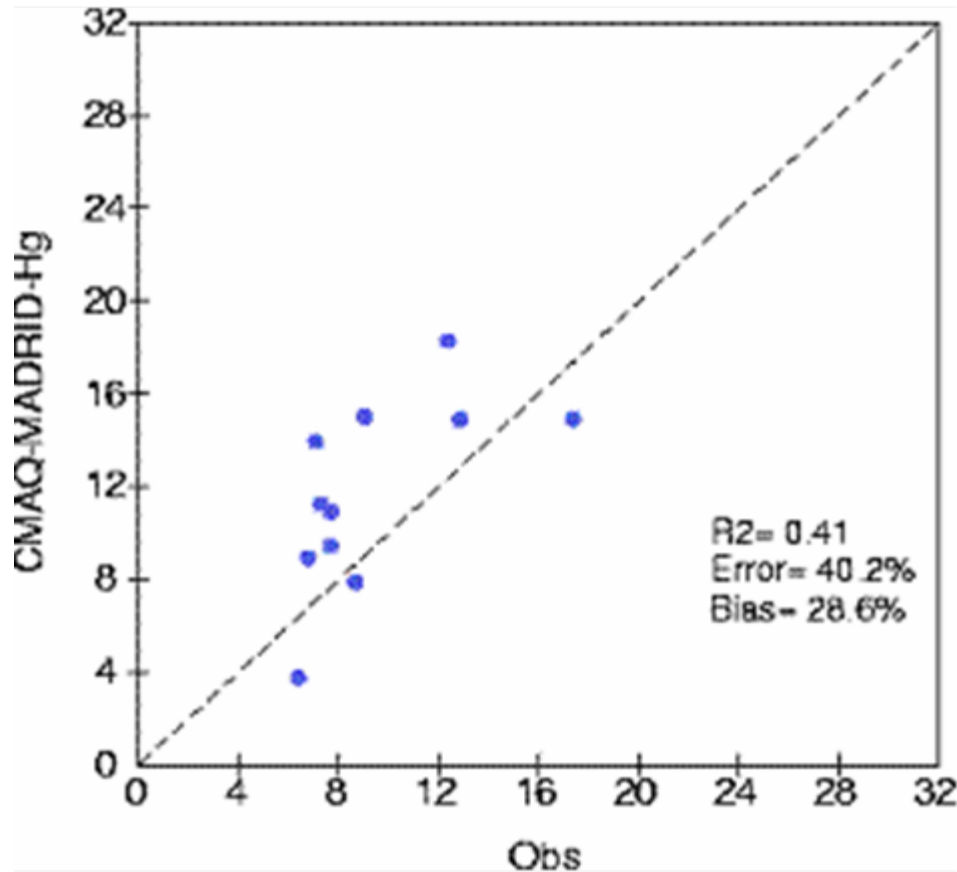
# Comparison of CMAQ-MADRID-Hg with 1996 MDN Data ( $\mu\text{g}/\text{m}^2\text{-yr}$ )





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# Comparison of CMAQ-MADRID-Hg with 1996 MDN Data ( $\mu\text{g}/\text{m}^2\text{-yr}$ )

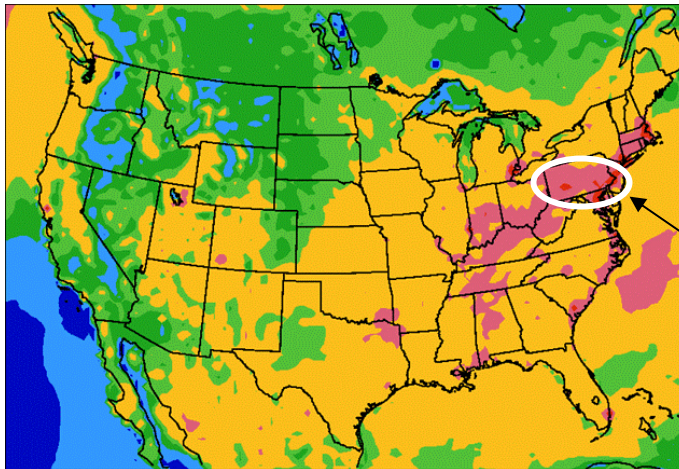


Mercury deposition was adjusted based on actual precipitation amounts

# Next Steps

## 1. Better treatment of plumes from elevated point sources

- Incorporate plume treatment within the 3-D grid model CMAQ-MADRID-Hg
  - More realistic representation of local impacts
  - Treatment of plume chemistry
  - Minimization of the “Pennsylvania anomaly”

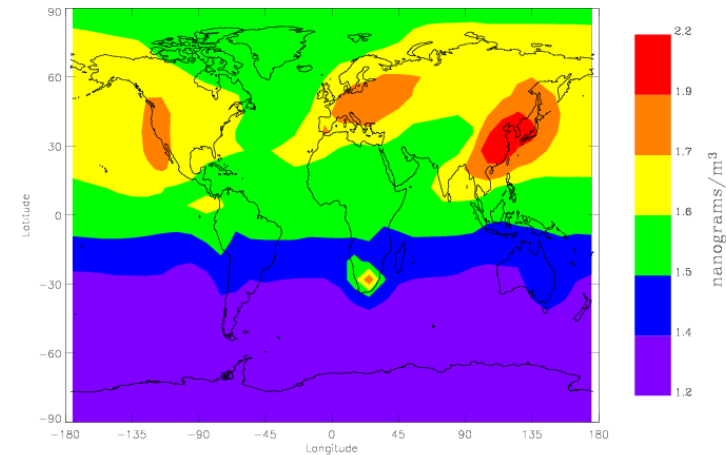


Overestimation of Hg deposition downwind of the Ohio Valley



# Power Plant Plume Mercury Chemistry

- Evidence of  $\text{Hg}^{\text{II}}$  reduction in power plant plumes
- Reduction of  $\text{Hg}(\text{II})$  by  $\text{SO}_2$  (possibly via heterogeneous reaction on particles) is compatible with the global Hg cycling budget
- Collaboration with Susannah Scott (UCSB) regarding laboratory experiments of plume chemistry

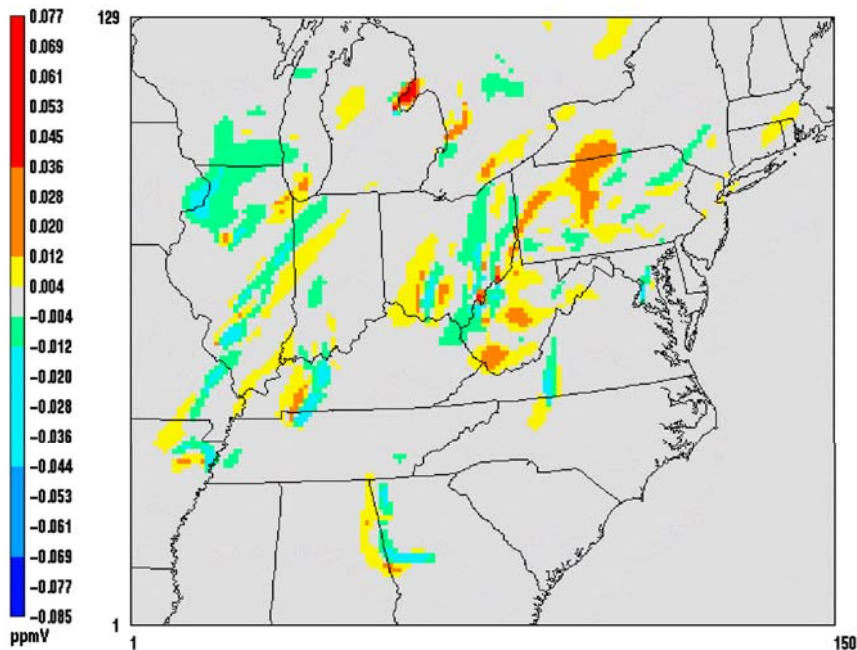




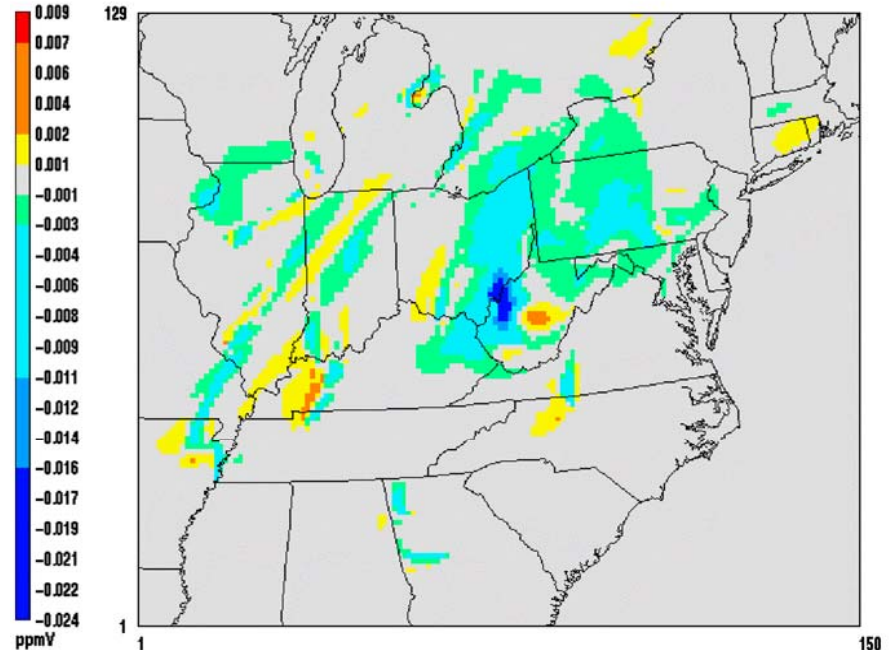
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# Plume-in-grid Treatment (APT) Effect on ozone and nitric acid

Differences between APT and the base CMAQ  
(40 power plants, July 1995 simulation, eastern United States)



Differences in Ozone Concentrations



Differences in HNO<sub>3</sub> Concentrations

One may expect significant effects for mercury as well



## 2. Mercury Model Intercomparison

- Use of common inputs
  - Meteorology
  - Emissions
  - Boundary conditions from a global model
- EPA Models
  - CMAQ-Hg
  - REMSAD
- AER/EPRI Models
  - CMAQ-MADRID-Hg (with and without plume treatment)
  - TEAM

# Conclusions

- CMAQ-MADRID-Mercury: “One-atmosphere” mercury model developed using CMAQ as the host model
- State-of-the-science treatment of PM (MADRID) and mercury processes
- Boundary conditions from a global mercury chemistry transport model
- Initial application to continental U.S. for 1996
- Subsequent steps:
  - Plume-in-grid treatment of mercury and power plant plume mercury chemistry
  - EPA’s mercury inter-comparison study

# Acknowledgements

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- EPRI



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